### BIOSYNTHESIS OF CANNABINOID ACIDS\*

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Abstract—Malonic acid, mevalonic acid, geraniol and nerol were incorporated into tetrahydrocannabinolic acid and cannabichromenic acid in *Cannabis sativa*. The pathway from cannabigerolic acid to tetrahydrocannabinolic acid via cannabidiolic acid was established by feeding labelled cannabinoid acids. Cannabichromenic acid was shown to be formed on a side pathway from cannabigerolic acid.

#### INTRODUCTION

Mechoulam and Gaoni have postulated the two parallel routes for the biogenesis of the cannabinoids [1], the first involves the condensation of geraniol with olivetol to give the neutral cannabinoids and the second, the condensation of geraniol with olivetolic acid to give the cannabinoid acids. In the course of our investigation on Cannabis [2], the amounts of the neutral cannabinoids were observed to increase in stored material, whereas several cannabinoid acids, including cannabigerolic acid (1), cannabidiolic acid (2), cannabichromenic acid (3) and tetrahydrocannabinolic acid (4) (Fig. 1), have been isolated from fresh plants. Previously, we noted that the neutral cannabinoids as well as cannabinolic acid are artifacts formed during harvesting and the storage, and proposed a biogenetic pathway of the cannabinoid acids [3].

This paper deals with feeding studies with *Cannabis* plants using <sup>14</sup>C-labelled malonate and mevalonate, <sup>3</sup>H-labelled geraniol and nerol, and with labelled cannabigerolic and cannabidiolic

acids in order to throw light on the overall biosynthetic pathway.

### RESULTS AND DISCUSSION

Strains of Cannabis sativa

For the feeding experiments, four strains of Cannabis sativa L., regarded as genetic types, were used (Table 1). Three of them, Mexican, Minamioshihara-1 (M-1) and Domestic-1 (D-1), have tetrahydrocannabinolic acid (4) as the major cannabinoid acid. M-1 also contains cannabigerolic acid monomethyl ether (5) as the second major constituent, Domestic-2 (D-2) contains cannabidiolic acid (2), but almost no 4, and thus seems to lack the capacity for production of the latter acid.

# Tracer compounds

The tracers employed in this study are shown in Table 2. Geraniol-[1-³H] and nerol-[1-³H] were prepared by NaB³H<sub>4</sub> reduction of the corresponding aldehydes. The labelled cannabinoid acids [U-¹<sup>4</sup>C] 1 and 2 were obtained from *Cannabis* plants cultivated with ¹<sup>4</sup>CO<sub>2</sub> [4]. Cannabinoid acids 1 and 2, labelled at the carboxyl carbon, were synthesized by the reaction of the neutral cannabinoids with ¹<sup>4</sup>C-labelled methyl mag-

<sup>\*</sup>This forms Part 8 of "Cannabis". Part 7: Shoyama, Y., Kuboe, K., Nishioka, I. and Yamauchi, T. (1972) *Chem. Pharm. Bull.* (Tokyo) **20**, 2072. This work was presented at the 88th(1968), 89th(1969), 91st(1971) and 92nd(1972) Annual Meetings of Pharmaceutical Society of Japan.

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Mevalonate

Acetate - Malonate

OH

OH

COOH

$$C_{12}$$
 - Polyketide

OH

 $C_{5}H_{11}$ 

OH

 $C_{5}H_{11}$ 

OH

 $C_{5}H_{11}$ 

OH

 $C_{5}H_{11}$ 

OH

 $C_{5}H_{11}$ 

OH

 $C_{5}H_{11}$ 
 $C_{5}H_{11}$ 

Fig. 1. Biosynthetic pathway of cannabinoid acids.

nesium carbonate [5]. Cannabigerolic acid-[U-<sup>3</sup>H] was prepared according to Wilzbach's procedure. The labelled substrates were fed to the plants either by using Bennett and Heftmann's procedure [6] (geraniol and nerol) or by means of cotton wicks. The final identification of the labelled cannabinoids isolated from plants is given in the Experimental.

## Incorporation of primary precursors

The biosynthesis of the cannabinoid acids from acetate-malonate (MA) and mevalonate (MVA) was shown by the incorporation of MA-[2-<sup>14</sup>C] and MVA-[2-<sup>14</sup>C] into 3 and 4 (Table 2). The incorporation ratio of both precursors, however, was quite low. Geraniol-[1-<sup>3</sup>H] and nerol-[1-<sup>3</sup>H] were administered, and the radioactive cannabinoid acids (3 and 4) were isolated and decarboxylated to the corresponding neutral compounds

(tetrahydrocannabinol and cannabichromene) in a chromatographycally pure form. The production of radioinactive cannabinol from the active tetrahydrocannabinol by the removal of C-1- $^{3}$ H [7], originally from geraniol- and nerol-1- $^{3}$ H, indicates the intact incorporation of both monoterpenes into **4**. Since **1** is formed from geraniol and C<sub>12</sub>-polyketide, some difference might have been

Table 1. Cannabinoid content of various strains of Cannabis sativa

	Canna	binoid	Preser	nt (%)*
Strain	2	3	4	5
Mexican	neg	0.58	2.10	neg
Minamioshihara No. 1 (M-1)	neg	0.17	0.76	0.18
Domestic No. 1†, (D-1)	neg	0.17	0.26	trace
Domestic No. 2 (D-2)	0.51	0.03	neg	trace

<sup>\*</sup> Contained in fresh dried leaves.

<sup>† 50-</sup>day-old plant.

Expt. no	Strain	Precursor Fed.	Incor. (%)*	Sp. act. (dpm/M)
1	Mexican	Malonic acid-[2- <sup>14</sup> C]-Ca (30 µCi)	4* 0·012 3* 0·007	
2	Mexican	( $\pm$ )-Mevalonic acid- [2- <sup>14</sup> C]-Na (20 $\mu$ Ci)	<b>4*</b> 0.013 <b>3*</b> 0.003	
3	Mexican	Geraniol-[1- $^{3}$ H] (1.67 × 10 $^{9}$ cpm)	<b>4*</b> 0.419 <b>3*</b> 0.044	11·1† 9·78†
4	Mexican	Nerol-[1- $^{3}$ H] (3.66 × 10 $^{8}$ cpm)	<b>4*</b> 0·284 <b>3*</b> 0·019	1·09† 0·748†
5	(D-2)	$1-[carboxyl-^{14}C]$ (3·15 × 10 <sup>6</sup> dpm)	<b>2</b> 3·16	1.32
6	Mexican	$1-[carboxyl-^{14}C]$ (3·15 × 10 <sup>6</sup> dpm)	<b>4</b> 2.94 <b>3</b> 0.12	0·826 0·249
7	(D-1)	$1-[carboxyl-^{14}C]$ (3·12 × 10 <sup>6</sup> dpm)	<b>4</b> 0.68 <b>3</b> 0.22	0·173 0·134
8	(M-1)	$1-[U-^3H]$ (1·22 × 10 <sup>9</sup> dpm)	<b>4*</b> 0·039 <b>3*</b> 0·023 <b>5*</b> 0·004	91·3 307·0 7·33
9	Mexican	$\frac{2}{(9.87 \times 10^5 \text{ dpm})}$	<b>4*</b> 0.42 <b>3*</b> 0	0·062 0
10	Mexican	2-[carboxyl- $^{14}$ C] (1.18 × 10 <sup>6</sup> dpm)	4 2.80	0.776

Table 2. Tracer experiments with various strains of C. sativa

expected between the incorporation ratios of geraniol and of nerol into 3 and 4, if they are biosynthesized through 1. However, no remarkable difference between the ratios was observed (Table 2, experiments 3 and 4), probably due to the equilibration of the terpenes before condensation with the olivetolic acid moiety. Similar results have been reported for some indole alkaloids [8].

Incorporation of cannabigerolic and cannabidiolic acids

When labelled 1 was fed, the label in the resulting metabolites varied according to the *Cannabis* variety to which it was applied. High incorporation into 2 was observed in variety D-2 (Expt. 5 Table 2), while 3 and 4 were labelled in Mexican and D-1 (Expts. 6 and 7). Upon application to M-1, 5 was observed to be labelled as well as 3 and 4 (Expt. 8). The decarboxylation of the acids furnished the corresponding neutral cannabinoids and radioactive CO<sub>2</sub>, indicating the direct transformation of 1 to 3 and 4. Cannabigerolic acid (1) was, therefore regarded as the first intermediate produced (Fig. 1).

Predominate incorporation of 1 in 3 as compared with that in 4 was observed mainly in young plants. These results are in good accord-

ance with the evidence previously reported by us [9] that 3 is present exclusively in young plants and is gradually replaced by 4 as the plants produce flowering shoots.

Upon administration of either cannabidiolic acid-[U-<sup>14</sup>C] or -[carboxyl-<sup>14</sup>C], to Mexican strain in which 3 was detected as a second major cannabinoid acid besides 4, the incorporation of 2 into the former substance was not observed and only the latter was labelled (Table 2 Expt. 9). When the carboxyl-<sup>14</sup>C acid was applied, the location of the radioactivity in the carboxyl moiety of the acids was determined in the same way as with cannabigerolic acid-[carboxyl-<sup>14</sup>C].

According to these feeding experiments, therefore, the biosynthetic pathway of the cannabinoid acids appears to be as shown in Fig. 1.

#### **EXPERIMENTAL**

Identification of neutral cannabinoids and cannabinoid acids. The following TLC [on Si gel G (Merck)] solvent systems were used:  $C_6H_6$  or hexane– $C_6H_6$ – $Et_2NH$  (20:10:1) [10] (for neutral cannabinoids); hexane–EtOAc (1:1) and  $C_6H_6$ –MeOH–HOAc (45:8:4) (for cannabinoid acids). Spots were detected with UV-light or with diazotized benzidine. Isolation of the labelled cannabinoids from the plants was effected using authentic crystalline cannabigerolic acid [1: mp 106–110°, UV $_{\rm max}^{\rm MOM}$  223 (20400), 267 (7400), 306 (2900). IR $_{\rm max}^{\rm CHCI_3}$  3320–3400

<sup>\*</sup> Compounds marked thus, isolated after decarboxylation as corresponding neutral compounds.

<sup>†</sup> Sp. act. marked thus, indicated by cpm/mM.

(OH), 1640 (C=O)] and the derivatives, tetrahydrocannabinol-3,5-dinitrophenylurethane [11] (mp 115-116°) and cannabidiolic acid diacetate [12] (mp 118-125°) in order to dilute labelled compounds and crystallize them to constant specific activity. Purity of uncrystallized cannabinoids (3, 4 and cannabichromene) were determined radiochromatographically.

Preparation of the labelled compounds. Geraniol-[1-3H] and nerol-[1-3H]: To a soln of geranial or neral in MeOH, NaB<sup>3</sup>H<sub>4</sub> (50  $\mu$ Ci) was added and the soln allowed to stand for 1 hr, dil with H<sub>2</sub>O, and extracted with Et<sub>2</sub>O. The extract was purified by preparative TLC to give radiochemically pure geraniol- and nerol-[1-3H]  $(2.01 \times 10^{10} \text{ cpm/mM} \text{ and } 4.55 \times 10^{10} \text{ cpm/mM})$ 10° cpm/mM, respectively). I-[carboxyl-14C] and 2-[carboxyl-14C]: Cannabigerol and cannabidiol respectively were treated with 14C-labelled methyl magnesium carbonate (produced from 3 mCi Ba<sup>14</sup>CO<sub>3</sub>) giving the cannabinoid acids which were identified by mmp, TLC, and GLC with authentic samples [1-(carboxyl-<sup>14</sup>C)], mp 106-110°, 2·08 × 10° dpm/mM, 2-[carboxyl-<sup>14</sup>C], mp of diacetate 118-125°, 1·83 × 10° dpm/mM). 1-[U-<sup>3</sup>H] was prepared according to William (10° 5.15° × 10° 2 dpm/mM). bach's procedure. mp  $106-110^{\circ}$ ,  $5.15 \times 10^{12} \text{ dpm/mM}$ . **2**-[U-<sup>14</sup>C]: Cuttings of variety D-2 (15 cm high, 20 leaves) were cultivated with 14CO<sub>2</sub> (10 mCi) for 6 days. Leaves were extracted with MeOH and the MeOH extractives were fractionated by Si gel (column and preparative TLC with hexane-EtOAc). 2-[U- $^{14}$ C], mp of the diacetate  $118-125^{\circ}$ ,  $5.07 \times$  $10^8 \, \text{dpm/mM}$ .

Feeding experiments. Female plants were used. The aq soln of MA-[2- $^{14}$ C]-Ca(30  $\mu$ Ci) and (±)-MVA-[2- $^{14}$ C]-Na(20  $\mu$ Ci) were fed to cuttings of Mexican variety (20 cm high, with 10 leaves) by means of cotton wicks. Geraniol-[1- $^{3}$ H] (1-67  $\times$  10° cpm) and nerol-[1- $^{3}$ H] (3-66  $\times$  108 cpm) were dissolved in Me<sub>2</sub>CO and applied on the leaves of Mexican variety (100 cm high, with 30 leaves) for 6 days according to the procedure of Bennett and Heftmann [6]. The labelled cannabinoid acids (105-106 dpm) were emulsified in H<sub>2</sub>O with the aid of Tween 85 and applied to the cuttings of the various strains shown in Table 2.

Isolation of the cannabinoids. The leaves were extracted with MeOH and the MeOH extractives were fractionated on a Si gel column (hexane-EtOAc) and preparative TLC (same solvent and C<sub>6</sub>H<sub>6</sub>-MeOH-HOAc). The cannabinoid acids isolated were then decarboxylated to neutral cannabinoids, which

were also obtained by direct decarboxylation of the total MeOH extractives, followed by the purification by preparative TLC.

Dehydrogenation. Tetrahydrocannabinol (19·4 mg, 7770 cpm; from geraniol-[1-3H] and 18·8 mg, 7740 cpm; from nerol-[1-3H]) was refluxed in toluene containing chloranil for 8 hr and purified by preparative TLC to give non-radioactive cannabinol.

Decarboxylation. Radioactive 3 (3·10  $\times$  10³ dpm, from Expt. 7) was decarboxylated by heating at 160° for 10 min to liberate radioactive CO<sub>2</sub> (2·95  $\times$  10³ dpm) which was trapped in the mixture of 2-methoxyethanol and ethanolamine (2:1) [13]. Radioactive 4 (8·24  $\times$  10⁴ dpm, from Expt. 10) and 2 (7·99  $\times$  10³ dpm, from Expt. 5) were decarboxylated in the same way to liberate radioactive CO<sub>2</sub> (6·18  $\times$  10⁴ dpm and 5·91  $\times$  10³ dpm, respectively).

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